

Supporting Information

Anisotropic Strain Tuning of L1₀ Ternary Nanoparticles for Oxygen Reduction

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Table S1. Reaction steps involved in dissociative mechanism (D₁-D₄) and associative mechanism (A₁-A₅).

<i>Mechanism</i>	<i>Reaction steps</i>
<i>Dissociative mechanism</i>	$\frac{1}{2} \text{O}_2 (\text{g}) + * \rightarrow \frac{1}{2} * \text{O}_2 (\text{D}_1)$ $\frac{1}{2} * \text{O}_2 (\text{g}) + * \rightarrow * \text{O} (\text{D}_2)$ $* \text{O} + \text{H}^+ + \text{e}^- \rightarrow * \text{OH} (\text{D}_3)$ $* \text{OH} + \text{H}^+ + \text{e}^- \rightarrow \text{H}_2\text{O}(\text{l}) + * (\text{D}_4)$
<i>Associative mechanism:</i>	$\text{O}_2 (\text{g}) + * \rightarrow * \text{O}_2 (\text{A}_1)$ $* \text{O}_2^* + \text{H}^+ + \text{e}^- \rightarrow * \text{OOH} (\text{A}_2)$ $* \text{OOH} + \text{H}^+ + \text{e}^- \rightarrow \text{H}_2\text{O}(\text{l}) + * \text{O} (\text{A}_3)$ $* \text{O} + \text{H}^+ + \text{e}^- \rightarrow * \text{OH} (\text{A}_4)$ $* \text{OH} + \text{H}^+ + \text{e}^- \rightarrow \text{H}_2\text{O}(\text{l}) + * (\text{A}_5)$

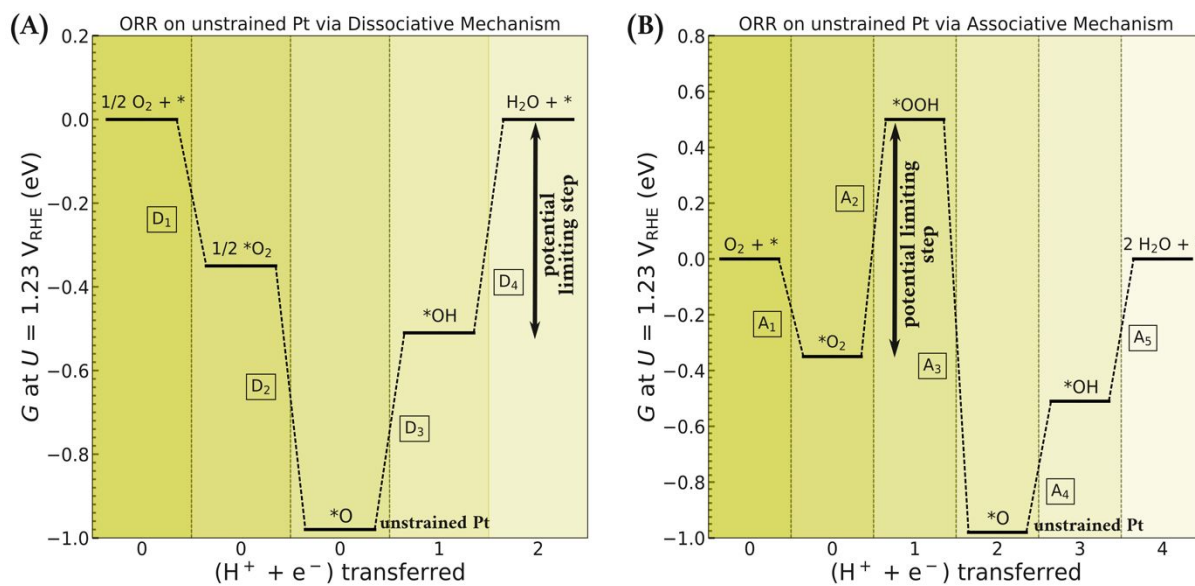


Figure S1. DFT-generated free energy landscapes at $U = 1.23 V_{\text{RHE}}$ for ORR on unstrained Pt (111) surfaces via (A) dissociative and (B) associative mechanism.

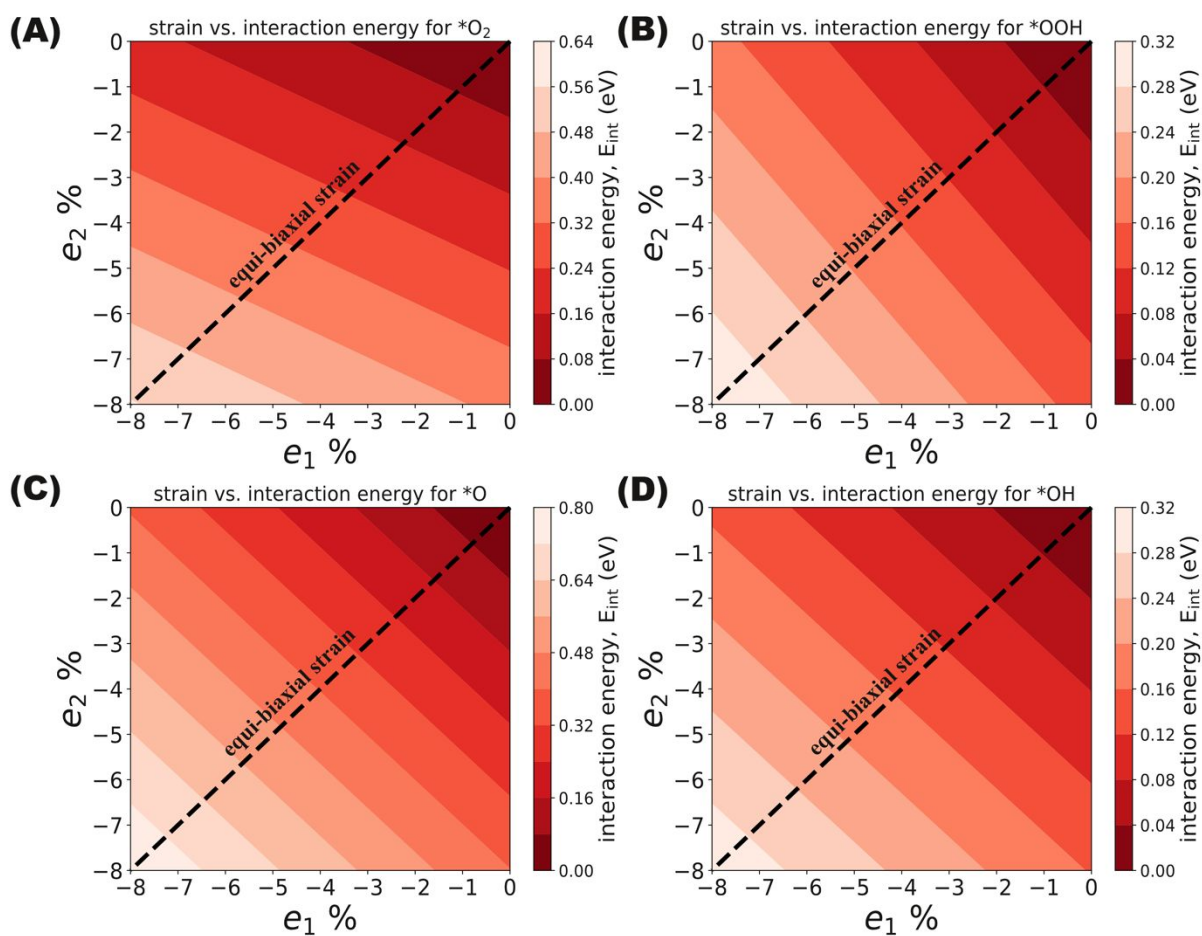


Figure S2. Eigenforce model-generated interaction energies as a function of 2D strain for (A) *O₂, (B) *OOH, (C) *O and (D) *OH.

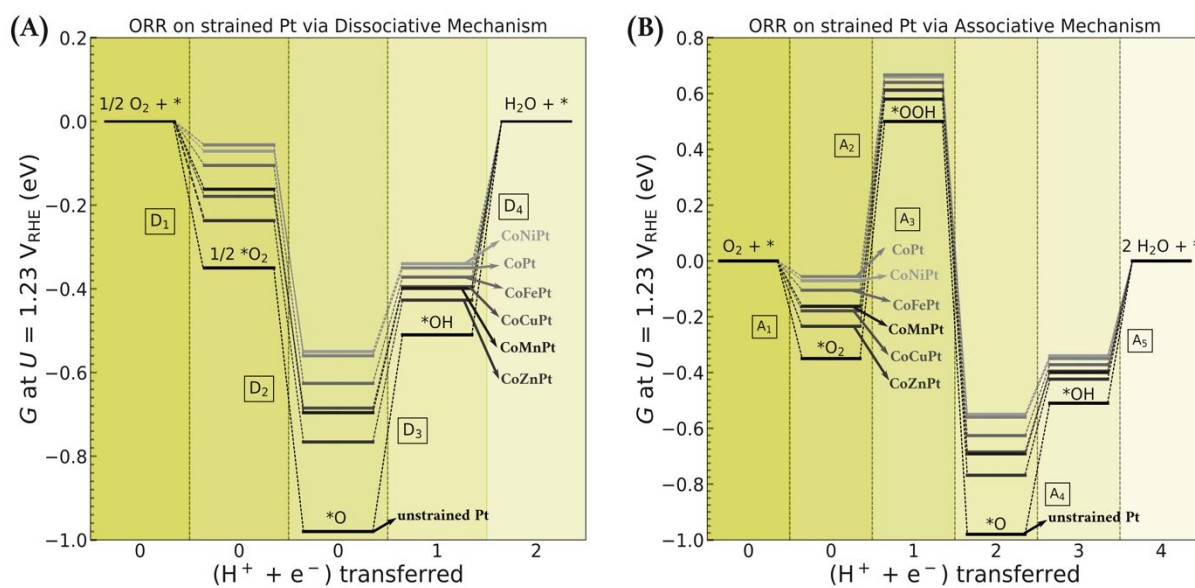


Figure S3. Computationally generated free energy landscapes at $U = 1.23 \text{ V}_{\text{RHE}}$ for ORR on L1₀-strained and unstrained Pt (111) surfaces via (A) dissociative and (B) associative mechanisms. The unstrained Pt surface data was generated using DFT calculations, while the eigenforce model was used for L1₀-strained Pt surfaces.

Table S2. Calculated change in free energies for the steps involved in the dissociative mechanism on strained L1₀ and unstrained fcc Pt (111) surfaces. Note that the values obtained for unstrained Pt are through DFT calculations while the eigenforce model was used for the strained surfaces. The step highlighted in bold denotes the potential-limiting step.

System	D ₁ (eV)	D ₂ (eV)	D ₃ (eV)	D ₄ (eV)
Unstrained Pt	-0.350	-0.630	0.470	0.510
L1 ₀ -CoNiPt	-0.071	-0.484	0.21	0.342
L1 ₀ -CoPt	-0.056	-0.503	0.213	0.349
L1 ₀ -CoFePt	-0.105	-0.520	0.254	0.372
L1 ₀ -CoCuPt	-0.179	-0.505	0.289	0.395
L1 ₀ -CoMnPt	-0.162	-0.533	0.296	0.399
L1 ₀ -CoZnPt	-0.237	-0.529	0.339	0.427

Table S3. Calculated change in free energies for the steps involved in the associative mechanism on strained L1₀ and unstrained fcc Pt (111) surfaces. Note that the values obtained for unstrained Pt are through DFT calculations while the eigenforce model was used for the strained surfaces. The step highlighted in bold denotes the potential-limiting step.

System	A ₁ (eV)	A ₂ (eV)	A ₃ (eV)	A ₄ (eV)	A ₅ (eV)
Unstrained Pt	-0.350	0.850	-1.480	0.470	0.510
L1 ₀ -CoNiPt	-0.071	0.737	-1.221	0.21	0.342
L1 ₀ -CoPt	-0.056	0.723	-1.226	0.213	0.349
L1 ₀ -CoFePt	-0.105	0.746	-1.266	0.254	0.372
L1 ₀ -CoCuPt	-0.179	0.792	-1.297	0.289	0.395
L1 ₀ -CoMnPt	-0.162	0.774	-1.307	0.296	0.399
L1 ₀ -CoZnPt	-0.237	0.817	-1.346	0.339	0.427

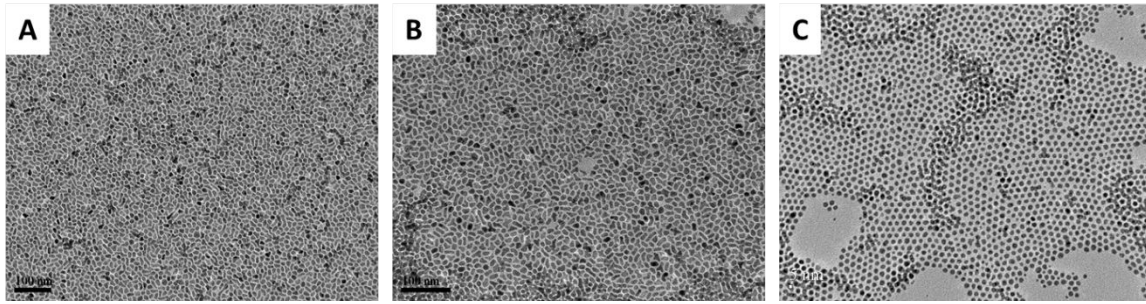


Figure S4. TEM images of as-synthesized (A) $\text{Co}_1\text{Ni}_2\text{Pt}_3$, (B) $\text{Co}_2\text{Ni}_1\text{Pt}_3$ and (C) Ni_1Pt_1 NPs.

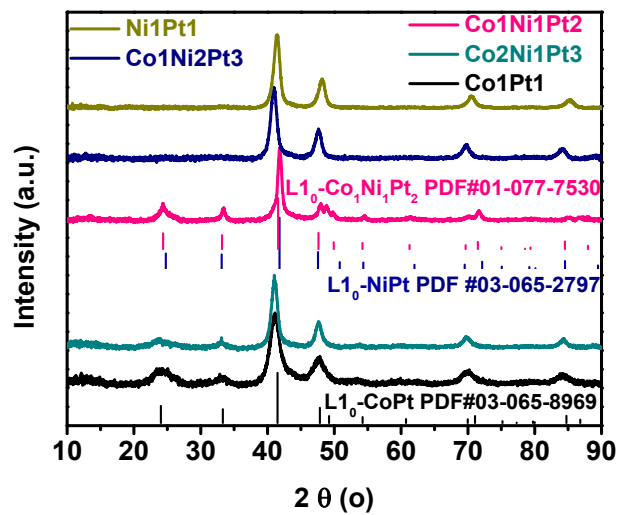


Figure S5. XRD patterns of Ni₁Pt₁, Co₁Ni₂Pt₃, Co₁Ni₁Pt₂, Co₂Ni₁Pt₃ and Co₁Pt₁ NPs annealed under 95 % Ar + 5% H₂ at 650 °C for 6 h.

Table S4. Composition ratio and Pt loading of C-loaded L1₀-NPs after post-acid treatment and annealing calculated from ICP results.

<i>Sample</i>	<i>composition</i>	<i>Pt loading (mg)</i>
L1 ₀ -CoPt	Co ₄₈ Pt ₅₂	11.7
L1 ₀ -CoMnPt	Co ₂₈ Mn ₂₀ Pt ₅₂	10.8
L1 ₀ -CoFePt	Co ₂₅ Fe ₃₀ Pt ₄₅	14.4
L1 ₀ -CoNiPt	Co ₂₉ Ni ₂₅ Pt ₄₆	12.6
L1 ₀ -CoCuPt	Co ₂₅ Cu ₂₇ Pt ₄₇	10.8
L1 ₀ -CoZnPt	Co ₁₉ Zn ₂₀ Pt ₆₁	12.1

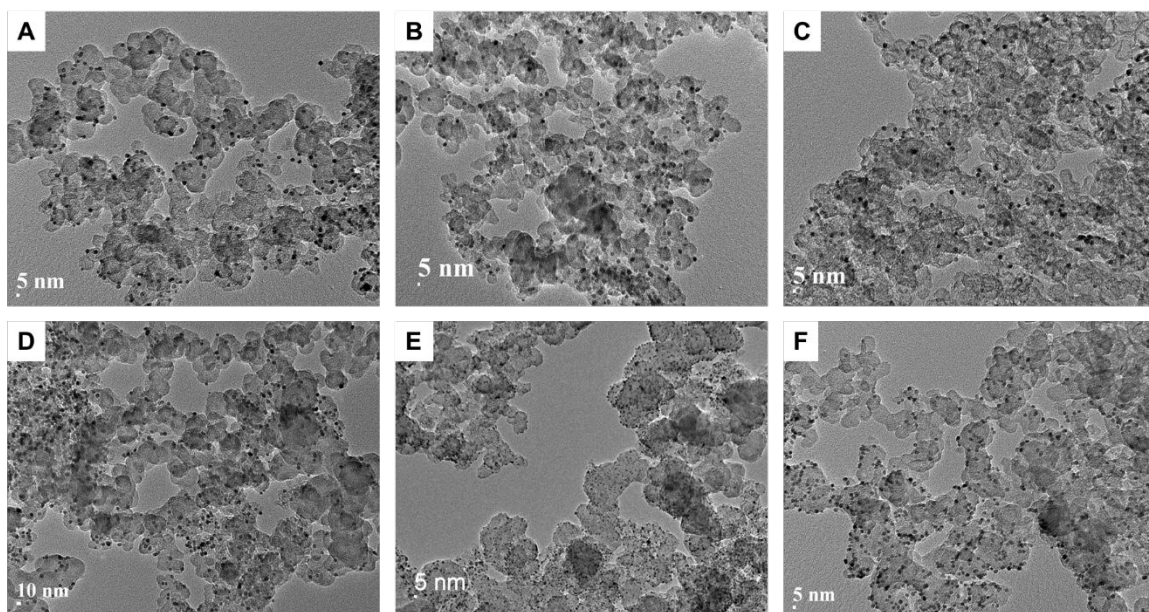


Figure S6. TEM images of C-loaded (A) L₁₀-CoPt NPs, (B) L₁₀-CoMnPt NPs, (C) L₁₀-CoFePt NPs, (D) L₁₀-CoNiPt NPs, (E) L₁₀-CoCuPt and (F) L₁₀-CoZnPt NPs after annealing treatment.

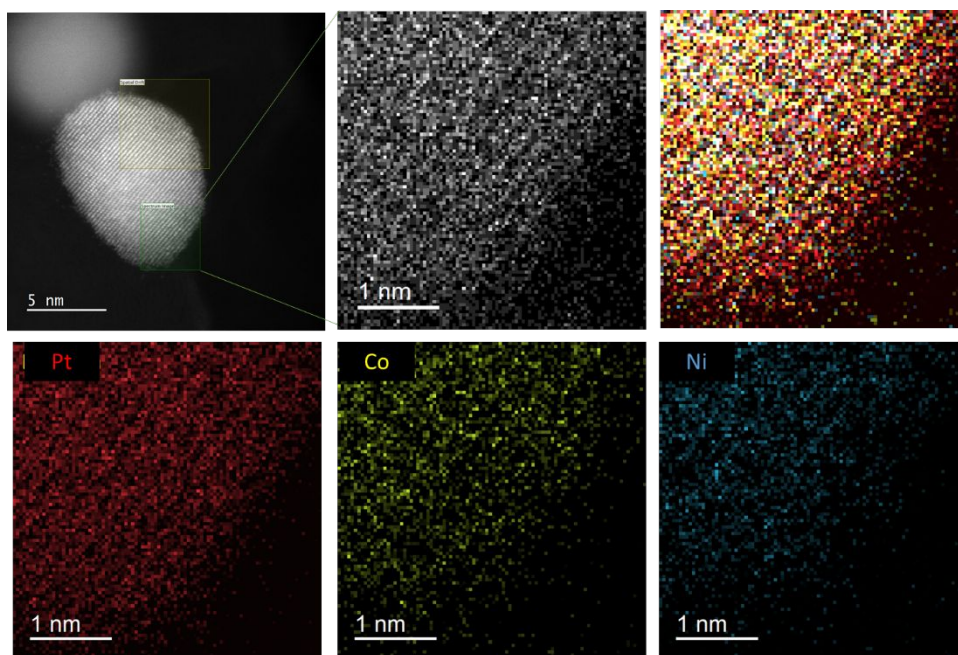


Figure S7. Atomic resolution elemental mapping images showing the alternative layers of Pt and Co/Ni.

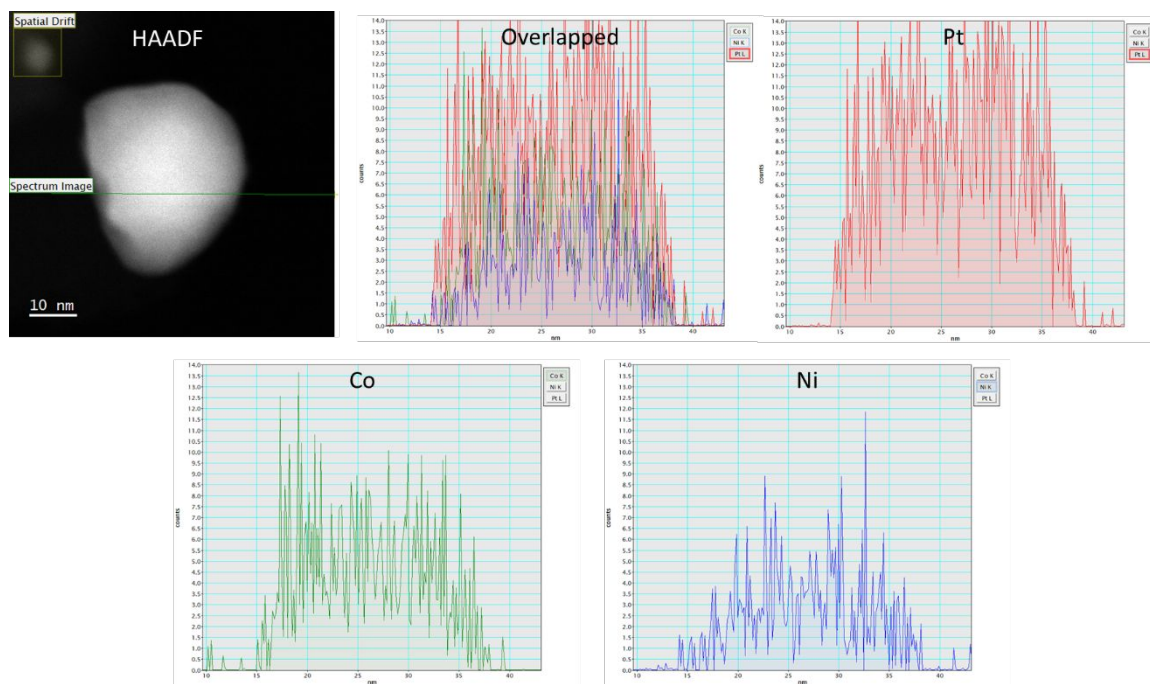


Figure S8. A HAADF image and a line-scan profile showing the distribution of Pt, Co and Ni components in core/shell structured L1₀-CoNiPt NPs.

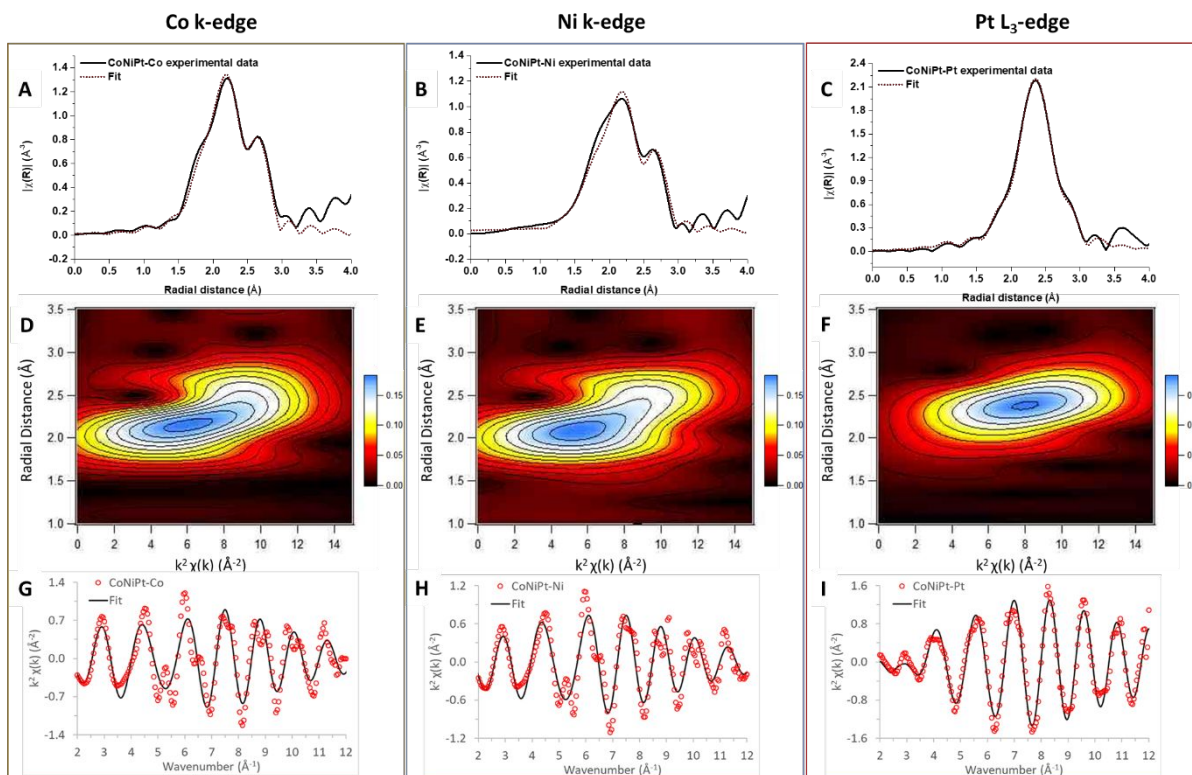


Figure S9. EXAFS experimental data and fitting results of Co K-edge, Ni K-edge and Pt L_{3} -edge. (A)-(C) FT-EXAFS spectra and best fit. (D)-(F) Contour map of k -space data. (G)-(I) k -space data and best fit.

Table S5 Fitting parameters, including coordination number (CN) and bond length (R), obtained from the fitting results of Pt L₃-edge EXAFS data.

<i>bond</i>	<i>CN</i>	<i>R (Å)</i>	$\sigma^2 (\text{Å}) \times 10^{-3}$	<i>E₀ (eV)</i>	<i>R factor</i>
Pt-Pt	6(2)	2.69(1)	4(2)	6(1)	0.0138
Pt-Co/Ni	5(2)	2.652(6)	6(3)	6(1)	0.0138

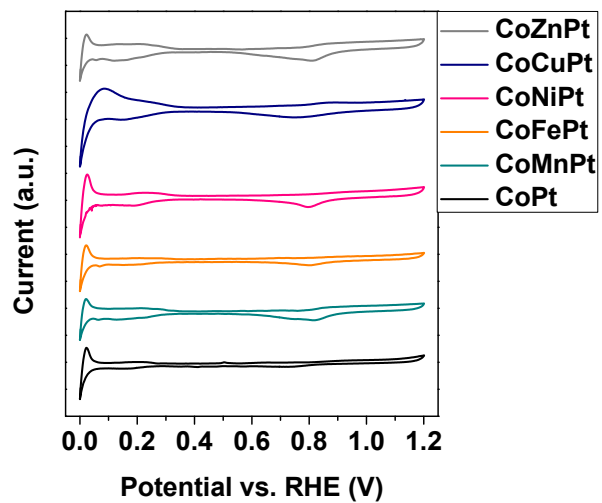


Figure S10. Cyclic voltammetry curves of C-L1₀-CoMPt NPs obtained in N₂-saturated 0.1 M HClO₄ at room temperature.

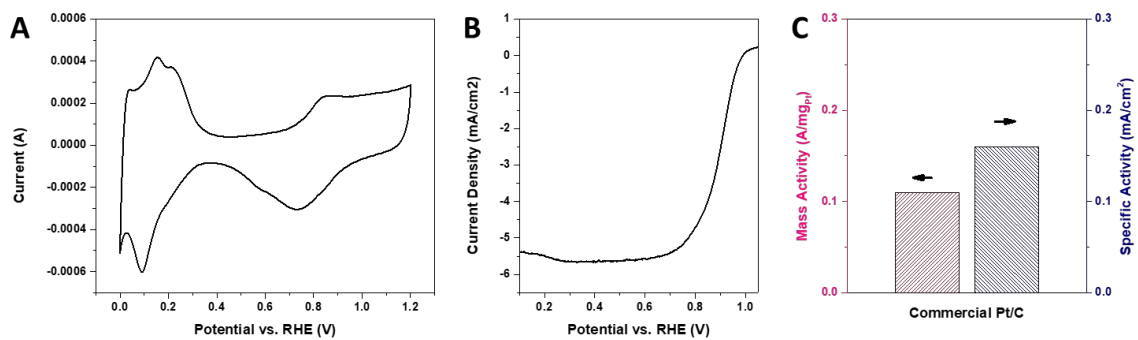


Figure S11. (A) Cyclic voltammetry curve of commercial Pt/C obtained in N_2 -saturated 0.1 M $HClO_4$ at room temperature. (B) Linear scanning voltammetry curve of commercial Pt/C collected in O_2 -saturated 0.1 M $HClO_4$ at room temperature. (C) Mass activity and specific activity calculated on commercial Pt/C.

Table S6. DFT optimized lattice constants using RPBE exchange-correlation functional.

System	a (Å)	b (Å)	c (Å)
Unstrained Pt	3.99	3.99	3.99
L1 ₀ -CoZnPt	3.97	3.97	3.70
L1 ₀ -CoMnPt	3.90	3.90	3.80
L1 ₀ -CoCuPt	3.94	3.94	3.66
L1 ₀ -CoFePt	3.86	3.86	3.82
L1 ₀ -CoPt	3.83	3.83	3.78
L1 ₀ -CoNiPt	3.87	3.87	3.69